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AN UNUSUAL SULFUR-CONTAINING DIKETOPIPERAZINE FROM THE BERMUDIAN SPONGE *TEDANIA IGNIS*¹

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ABSTRACT.—A new sulfur-containing diketopiperazine, cyclo(L-Pro-L-thioPro) [**1**], was isolated from extracts of the Bermudian sponge *Tedania ignis*. The structure was determined through spectral analyses and confirmed by synthesis.

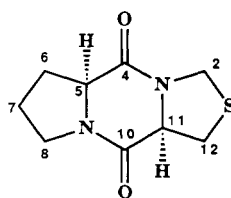
In the course of our investigation of the minor aromatic metabolites in extracts of the sponge *Tedania ignis* Duchassaing and Michelotti (Demospongiae) (1,2), we encountered numerous fractions containing diketopiperazines. Schmitz *et al.* (3) had previously reported several such constituents of *T. ignis*, and our laboratory has recently shown that the diketopiperazines reported by Schmitz's group are actually produced by an associated bacterium (4). Herein we report the structure of a novel sulfur-containing cyclic dipeptide from extracts of Bermudian *T. ignis*.

The collection, extraction, and fractionation of *T. ignis* have been described (2). Fractionation of both the CCl₄- and CHCl₃-soluble fractions of the organic extract by a series of gel permeation chromatographies (BioBeads S-X4 and Sephadex LH-20) gave mixtures of diketopiperazines which were resolved by centrifugal countercurrent chromatography [CHCl₃-MeOH-H₂O (25:34:20), ascending]. In addition to diketopiperazines reported earlier by Schmitz *et al.* (3), we found cyclo(L-Pro-L-Phe) and a new compound. Cyclo(L-Pro-L-Phe) was readily identified by spectral comparisons with literature data (5).

Accurate mass measurements (eims) provided the molecular formula

C₉H₁₂N₂O₂S for the new compound. The ¹³C-nmr spectrum contained resonances for two amide carbonyl groups and seven sp³ carbons (δ 23–63); the molecule had to be tricyclic. The absence of N-H stretching absorption in the ir and the lack of exchangeable protons in the ¹H nmr indicated that both amides were tertiary. The ¹H-nmr spectrum consisted of a series of resonances between 1.5 and 4.8 ppm which were well resolved at 500 MHz. The chemical shifts and multiplicities of certain signals (C/H-6,7,8; see Table 1) were characteristic of a proline fragment (1,2,5); this assignment was confirmed by ¹H-¹H COSY analyses. The proline and two tertiary amide residues left a C₃H₅S fragment and two sites of unsaturation to be assigned. An isolated methylene group (δ 4.75 and 4.50, ea 1 H, d, J = 10 Hz) had to be placed between a nitrogen and the sulfur; the azathioacetal carbon resonated at 48.5 ppm. The remaining elements comprised an ABX system. This new compound had to be **1**, a diketopiperazine constructed from the common amino acid proline and the relatively rare 3-thioprolinone.

The absolute stereochemistry was defined as *S,S* by synthesis from the com-



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TABLE 1. Nmr Assignments for
 Cyclo(L-Pro-L-thioPro) [1].^a

Position	¹³ C	¹ H
2	48.5	4.75, 4.50, ea 1H, d, 10
4	165.8 ^b	—
5	60.5	4.20, 1H, t, 8
6	27.8	2.15, 2.35, ea 1H, m
7	23.2	1.95, 2.05, ea 1H, m
8	45.3	3.55, 2H, m
10	164.2 ^b	—
11	62.8	4.45, 1H, t, 7
12	32.5	3.45, 3.35, ea 1H, dd, 12, 7

^aRecorded in CDCl₃; assignments based on ¹H-¹H COSY and ¹H-¹³C HETCOR experiments. ¹³C-nmr entries are δ; ¹H-nmr entries are δ, multiplicity, *J* in Hz.

^bAssignments may be interchanged.

mercially available L-amino acids. Using the procedures of Itoh *et al.* (6) and Nitecki *et al.* (7), the methyl ester of proline was coupled with the *t*-BOC derivative of 3-thioprolinone. The protecting groups were then removed under hydrolytic conditions, and the second amide bond was formed. This approach is known to proceed without racemization. The synthetic material, [α]_D - 112.8°, was identical in all respects with the natural product, [α]_D - 114.3°. Compound **1** was, therefore, cyclo(L-Pro-L-thioPro).

While **1**, a novel compound, was inactive in all our in-house bioassays (brine shrimp cytotoxicity, phytotoxicity, plant growth regulatory, antimicrobial, and insecticidal), other thiazolidine-carboxylic acid (3-thioprolinone) derived diketopiperazines have been prepared and found to have mild to strong neurosedative effects (8).

EXPERIMENTAL

ISOLATION.—The collection, extraction, and early fractionation steps have been described elsewhere (2). Compound **1** was isolated from the CCl₄ and CHCl₃ solubles by a sequence of gel permeation chromatographies: BioBeads S-X4 (fraction 7), Sephadex LH-20 (fractions 1 and 2), and a second Sephadex LH-20 (fraction 3). Final purification employed centrifugal countercurrent chromatography [CHCl₃-MeOH-H₂O (25:34:20), ascending].

CHARACTERIZATION.—Yield 4.8 mg (from

86 g crude extract); [α]_D²³ - 114.3° (c = 0.14, CHCl₃); ν max (CDCl₃) 1659, 1423 cm⁻¹; ms *m/z* (%) [M]⁺ 212.0648 (100) (C₉H₁₂N₂O₂S requires 212.0620), 138 (17), 124 (20), 87 (15); ¹H and ¹³C nmr see Table 1.

SYNTHESIS OF CYCLO(L-PRO-L-THIOPRO)
[1].—To 0.133 g (1.0 mM) of L-thioprolinone was added 0.2735 g (1.1 mM) BOC-ON, 10 ml of 15% aqueous Me₂CO, and 200 μl triethylamine. The mixture was stirred for 2 h at room temperature. The Me₂CO was evaporated, and the remaining aqueous phase was washed with EtOAc. The aqueous layer was then acidified with 1 M HCl and extracted with CH₂Cl₂. Evaporation of the CH₂Cl₂ phase gave the crude product, 0.1792 g (64%).

The BOC-thioprolinone (0.1792 g) was dissolved in 10 ml CH₂Cl₂ and 140 μl triethylamine. L-Proline methyl ester hydrochloride (0.1648 g, 1.0 mM) was added along with 0.1920 g (1 mM) *N*-ethyl-*N'*-(3-dimethyl-aminopropyl)-carbodiimide hydrochloride. The mixture was stirred overnight at -5°. The solution was washed sequentially with equal volumes of H₂O, 1 N citric acid, 5% NaHCO₃, and H₂O. The organic layer, which contained the crude product, was evaporated to dryness.

The crude *t*-BOC-dipeptide methyl ester was dissolved in 10 ml of HCO₂H and stirred at room temperature for 2 h. The HCO₂H was removed in vacuo, and the dipeptide ester formate was dissolved in 5 ml toluene and 25 ml *sec*-butanol. The mixture was refluxed for 2 h and the solvents removed by rotary evaporation.

The crude diketopiperazine was purified on a Sephadex LH-20 column (158 × 2.0 cm) using MeOH-MeCN (4:1). Evaporation of fraction 5 gave a white crystalline solid (63%) whose spectral data matched that of the natural product: [α]_D²³ - 112.8° (c = 0.19, CHCl₃); mp 168–170°.

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LITERATURE CITED

- R.L. Dillman, "Secondary Metabolites from the Bermudian Sponge *Tedania ignis*," Ph.D. Dissertation, Montana State University, Bozeman, MT 1990.
- R.L. Dillman and J.H. Cardellina II, *J. Nat. Prod.*, **54**, 1056 (1991).
- F.J. Schmitz, D.J. Vanderah, K.H. Hollenbeck, C.E.L. Enwall, Y. Gopichand, P.K. SenGupta, M.B. Hossain, and D. van der Helm, *J. Org. Chem.*, **48**, 3941 (1983).

4. A.C. Stierle, J.H. Cardellina II, and F.L. Singleton, *Experientia*, **44**, 1021 (1988).
5. A.C. Stierle, J.H. Cardellina II, and G.A. Strobel, *Proc. Natl. Acad. Sci. USA*, **85**, 8008 (1988).
6. M. Itoh, D. Hagiwara, and T. Kamiya, *Tetrahedron Lett.*, 4393 (1975).
7. D.E. Nitecki, B. Halpern, and J.W. Westley, *J. Org. Chem.*, **33**, 864 (1967).
8. N. Margoum, P. Tronche, P. Bastide, J. Bastide, and C. Rubat, *Eur. J. Med. Chem.*, **19**, 415 (1984).

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